

2012/2013 sem I Env. Engg

1a) (i)  $(25000 \text{ m}^3/\text{d} \div 500000 \text{ people}) \times 1000 = 500 \text{ Lpcd}$

(ii) Arithmetic rate  $\rightarrow$  Assuming population grows linearly

Annual growth rate  $\rightarrow (80000 - 50000) / 20 = 1500 \text{ people/year}$

No. of people to fully utilize design capacity  $\rightarrow (37000 \text{ m}^3/\text{d} \div 500 \text{ Lpcd})$   
 $= 74000 \text{ people}$

b) we assume that the per capita water consumption remains the same

since  $P_t = P_0 + K \cdot t$

$= 74000 = 50000 + 1500t$

$\therefore t = 16 \text{ years}$

(iii) Yes, we can reduce water demand by implementing a water

conservation programme by installing dual-flush cisterns, water trimble and reducing leaking leakage in pipes and fittings

1b) 24 hr Pumping Rate : Pumping  $\rightarrow (26400 \div 24 \text{ hrs}) \times 2 = 2200 \text{ m}^3$

Time	Demand (m <sup>3</sup> )	Pumping	Excess Pumping	Excess Demand
12-2am	610		1590	
2-4am	600		1600	
4-6am	650		1550	
6-8am	1350		850	
8-10am	2500			300
10-12pm	2750			550
12-2pm	2950			750
2-4pm	3250	2200 m <sup>3</sup>	1050	
4-6pm	3000		800	
6-8pm	3800		1600	
8-10pm	3590		1390	
10-12am	1350		1850	

$26400 = 6440 \text{ m}^3$        $6440 \text{ m}^3 = 6440 \text{ m}^3 \rightarrow \text{Both same (check)}$

Equalising storage  $\rightarrow 6440 \text{ m}^3$

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(ii) minimum: \_\_\_\_\_

maximum: 4mm (

Equalising storage at 6pm  $\rightarrow 1800m^3$

(iii) 12-8pm. That's when the demand for water is the highest

2a) component concentration. Eqv. lwt. conc.

(i)	(mg/L)	(mg/meq)	(meq/L)
Ca <sup>2+</sup>	94	20	94/20 = 4.7
Mg <sup>2+</sup>	24	12.2	24/12.2 = 1.96
Na <sup>+</sup>	14	23	14/23 = 0.61
SO <sub>4</sub> <sup>2-</sup>	67	48	1.395
Cl <sup>-</sup>	24	35.5	0.676

Total cation (+) = 4.7 + 1.96 + 0.61

Total Anion (-) = 1.395 + 0.676 = 2.071 meq/L

It's incomplete as  $\Sigma$  Anion  $\neq$   $\Sigma$  cation

(ii) concentration (meq/L)  $\rightarrow 7.27 - 2.071 = 5.20$  meq/L

concentration (mg/L)  $\rightarrow 5.20$  meq/L  $\times$  61 mg/meq = 317.2 mg/L

(iii)

cations	Ca <sup>2+</sup> 4.7	Mg <sup>2+</sup> 1.96	Na <sup>+</sup> 0.61
anions	HCO <sub>3</sub> <sup>-</sup> 5.20	SO <sub>4</sub> <sup>2-</sup> 1.395	Cl <sup>-</sup> 0.676

Hypothetical combinations

Ca(HCO<sub>3</sub>)<sub>2</sub> = 4.7 meq/L / MgSO<sub>4</sub> = 1.395 meq/L

Mg(HCO<sub>3</sub>)<sub>2</sub> = 0.504 meq/L / MgCl<sub>2</sub> = 0.065 meq/L

NaCl = 0.61 meq/L

(iv) carbonate hardness  $\rightarrow 5.20$  meq/L  $\times$  50 mg/meq = 260 mg/L as CaCO<sub>3</sub>

Total hardness  $\rightarrow (4.7 + 1.96)$  meq/L  $\times$  50 mg/meq = 333 mg/L as CaCO<sub>3</sub>

Non-carbonate  $\rightarrow 333 - 260$  mg/L as CaCO<sub>3</sub> = 73 mg/L as CaCO<sub>3</sub>

2bi) vol. of tank  $\rightarrow 10 \times 3 \times 3 = 90 \text{ m}^3$

$Q = 4500 \text{ m}^3/\text{d}$

(i) Total power  $\rightarrow G^2 M V = P$

$P = (40)^2 (0.09 \times 10^{-3}) (90) = 128.16 \text{ W}$

(ii)  $N = 15 \text{ rpm} = 5/60 \text{ rps} = 0.0833 \text{ rps}$

$V_p = 2\pi r N = 2\pi(1)(0.0833 \text{ rps}) = 0.524 \text{ m/s}$  (For one paddle blade)

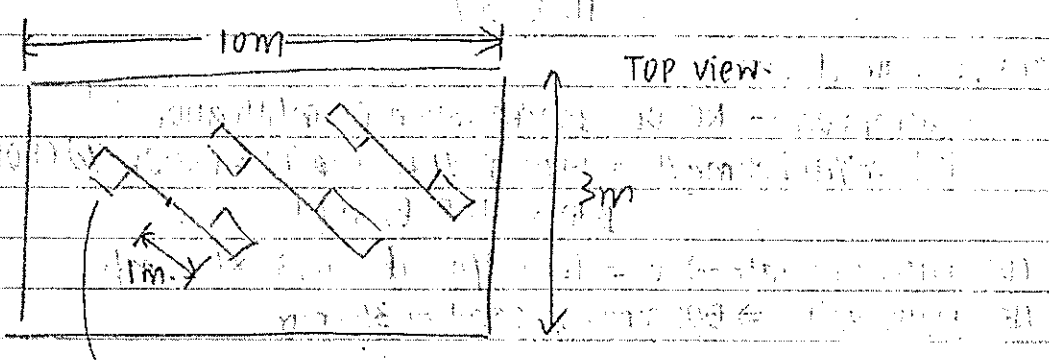
$P = 0.5 G v A \rho v^3$  (where  $v = 0.75 V_p$ )

$128.16 = 0.5(1.0)A(1000)(0.75(0.524))^3$

$A = 2.346 \text{ m}^2$  (Total Area)

Area of one paddle  $\rightarrow 2.346 \div (3 \text{ paddle wheel} \times 2 \text{ paddle blade})$

$= 0.391 \text{ m}^2$



Each Area =  $0.391 \text{ m}^2$

biii) HRT  $\rightarrow V/Q = 90 \text{ m}^3 / 4500 \text{ m}^3/\text{d} = 20 \text{ min}$ .  
 Significance of.

(1) Hydraulic detention time  $\rightarrow$  How long water stays in tank (HRT)

(2) mean velocity gradient  $\rightarrow$  (degree of mixing)

Together, they determine the total number of particle collision

3ai) Combined water  $\rightarrow 400 \text{ ha} \times 60 \text{ people/ha} \times 0.320 \text{ m}^3/\text{pcd} = 7680 \text{ m}^3/\text{d}$

$$+ 5 \text{ m}^3/\text{ha} \cdot \text{d} \times 400 \text{ ha} = 2000 \text{ m}^3/\text{d}$$

$$+ 80/1000 \text{ m}^3/\text{pcd} \times 60 \text{ people/ha} \times 400 \text{ ha} = 1920 \text{ m}^3/\text{d}$$

$$= 11600 \text{ m}^3/\text{d}$$

a ii) Combined BOD

Assumption  $\rightarrow$  NO BOD concentration in infiltration

$$7680 \text{ m}^3/\text{d} (250 \text{ mg/L}) + 1920 \text{ m}^3/\text{d} (200 \text{ mg/L}) = 11600 \text{ m}^3/\text{d} (\text{BOD})$$

$$\text{BOD} = 198.62 \text{ mg/L}$$

(b) Filtration rate  $\rightarrow v = 150 \text{ m}^3/\text{m}^2 \cdot \text{d} = 1.736 \times 10^{-3} \text{ m/s}$

(i) Anthracite  $\Rightarrow 500 \text{ mm}$ , sand  $\Rightarrow 300 \text{ mm}$ .

Anthracite

$$Re = \frac{\gamma d v}{\nu} = \frac{0.9 (1 \times 10^{-3}) (1.736 \times 10^{-3})}{1 \times 10^{-6}}$$

$$= 1.5624$$

$$C_D = \frac{24}{1.5624} + \frac{3}{\sqrt{1.5624}} + 0.34 = 18.11$$

$$h = \frac{1.067 (18.11) (1.736 \times 10^{-3})^2}{0.9 (9.81) (1 \times 10^{-3}) (0.45)^4}$$

$$= 80.42 \text{ mm}$$

sand

$$Re = \frac{0.75 (0.5 \times 10^{-3}) (1.736 \times 10^{-3})}{1 \times 10^{-6}}$$

$$= 0.651$$

$$C_D = \frac{24}{0.651} = 36.86$$

$$h = \frac{1.067 (36.86) (1.736 \times 10^{-3})^2}{0.75 (9.81) (0.5 \times 10^{-3}) (0.40)^4}$$

$$= 377.6 \text{ mm}$$

Total Head loss  $\rightarrow 377.6 + 80.42 = 458.02 \text{ mm}$

(ii) weighted  $f \rightarrow 0.45 (500) + 0.4 (300) = f_w (800)$

$$f_w = 0.43$$

$$1.4 = 1 - 0.43$$

$$\frac{1}{1 - f_e} \quad f_e = 0.592$$

$$V_b = 0.12 (0.592)^{4.5} = 0.011 \text{ m/s}$$

(iii)  $d_2 = \text{Anthracite}$ 

$$d_2 = 0.5 \times 10^{-3} \left( \frac{2.65 - 1}{1.55 - 1} \right)^{2/3} = 1.04 \times 10^{-3} \text{ m} = 1.04 \text{ mm} > 1 \text{ mm}$$

 $\therefore$  No intermixing(c)  $Q = 90000 \text{ m}^3/\text{d}$ 

$$Q \text{ for 1 tank} \rightarrow 90000 \text{ m}^3/\text{d} \div 4 = 22500 \text{ m}^3/\text{d}$$

$$V/Q = 3 \text{ hr} \rightarrow V / 22500 \text{ m}^3/\text{d} \times 24 = 3$$

$$V = 2812.5 \text{ m}^3$$

let width =  $w$ 

$$\therefore \text{Area} = 2.5w^2$$

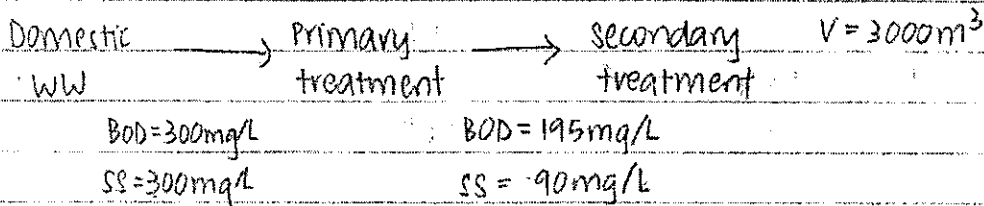
$$Q/A = \text{overflow rate} \rightarrow 22500 \text{ m}^3/\text{d} \div 2.5w^2 \text{ m}^2 = 35 \text{ m}^3/\text{m}^2 \cdot \text{d}$$

$$w = 16 \text{ m}$$

$$l = 16 \times 2.5 = 40 \text{ m}$$

$$\text{Depth} \rightarrow 2812.5 \text{ m}^3 \div (16 \times 40 \text{ m}^2) = 4.39 \text{ m} \approx 4.4 \text{ m}$$

4a)



$$(i) \text{ Aeration} \rightarrow V/Q = 3000 \text{ m}^3 \div (10000 \text{ m}^3/\text{d}) \times 24 \text{ hrs/d} = 7.2 \text{ hrs}$$

$$(ii) \text{ F/m ratio} \rightarrow \frac{Q \cdot S_0}{VX} = \frac{10000 \text{ m}^3/\text{d} (3000 \text{ mg/L})}{3000 \text{ m}^3 (3000 \text{ mg/L}) (0.8)} = 0.27$$

$$(iii) \text{ VLR} = \frac{Q \cdot S_0}{V} = \frac{10000 \text{ m}^3/\text{d} (3000 \text{ mg/L})}{3000 \text{ m}^3} = 0.65 \text{ kg/m}^3 \cdot \text{d}$$

(iv) mean cell residence time  $\rightarrow V_r X$ 

$$Q_e X_e + Q_w X_r \quad \text{since } X_e = 0 \therefore$$

$$\theta_c = \frac{3000 \text{ m}^3 (3000 \text{ mg/L}) (0.8)}{100 \text{ m}^3/\text{d} (1\% \times 10^6 \text{ mg/L}) (0.8)} = 9 \text{ days}$$

$$(v) \frac{Q_r}{Q} = \frac{X}{X_r - X} \quad \therefore Q_r = \frac{(10000)(3000)(0.8)}{(10000) - (3000)(0.8)} = 3157.9 \text{ m}^3/\text{d}$$

4b) Primary sludge  $\rightarrow$  Solids + water  
5% + 95% = 100%

(Primary) solids  $\rightarrow$  Volatile + Fixed  
60% + 40% = 100%

$$\frac{W_s}{P_s} = \frac{W_f}{P_f} + \frac{W_w}{P_w} \quad \begin{matrix} P_w = 1.0 \\ P_f = 2.8 \end{matrix}$$

$$1 = \frac{0.4}{2.8} + \frac{0.6}{1.0}$$

$$P_s = 1.34$$

$$\frac{W_{s1}}{P_{s1}} = \frac{W_w}{P_w} + \frac{W_s}{P_s}$$

$$1 = \frac{0.95}{1} + \frac{0.05}{1.34}$$

$$P_{s1} = 1.01$$

Wasted Activated Sludge  $\rightarrow$  Solids + water  
1% + 99% = 100%

Wasted (solids)  $\rightarrow$  Volatile + Fixed  
80% + 20% = 100%

$$\frac{W_s}{P_s} = \frac{W_f}{P_f} + \frac{W_w}{P_w} \quad , \quad P_w = 1.15$$

$$1 = \frac{0.2}{2.8} + \frac{0.8}{1.0}$$

$$\frac{W_{s1}}{P_{s1}} = \frac{W_w}{P_w} + \frac{W_s}{P_s}$$

$$1 = \frac{0.99}{1} + \frac{0.01}{1.15} \quad , \quad P_{s1} = 1.00$$

$$\text{Vol. of Pri sludge} \rightarrow \frac{(10000 \text{ m}^3/\text{d} \times 300 \text{ mg/L} \times 0.6)}{1.01(0.05)} = 35643 \text{ m}^3$$

$$\text{Vol. of wasted sludge} \rightarrow \frac{10000 \text{ mg/L} \times 100 \text{ m}^3/\text{d}}{1(0.01)} = 10000 \text{ m}^3$$

$$\text{Total sludge vol.} \rightarrow 35643 + 10000 = 45643 \text{ m}^3 \text{ daily}$$

4c) This is a design question.  $\therefore$  the answers you get might not necessarily be the same as mine as we might make different assumptions. (But the steps are rather similar).

effluent BOD (using mass balance)

$$\hookrightarrow 20000 \text{ m}^3/\text{d} (3 \text{ mg/L}) + 5000 \text{ m}^3/\text{d} (\text{Eff. BOD}) = 25000 \text{ m}^3/\text{d} (10 \text{ mg/L})$$

$$\text{Effluent BOD} \rightarrow 38 \text{ mg/L}$$

BOD after primary treatment.

$$\hookrightarrow 400 \text{ mg/L} \times (100\% - 35\%) = 260 \text{ mg/L} = S_0$$

$$\text{treatment efficiency by trickling filter} \rightarrow \frac{260 - 38}{260} = 85.4\%$$

Using NRC eqn

$$W = \frac{Q \cdot S_0}{V} = \frac{20000 \text{ m}^3/\text{d} (260 \text{ mg/L})}{V} = \frac{5200 \text{ kg BOD/d}}{V}$$

Assume recirculation ratio,  $R = 2 \rightarrow \frac{Q_r}{Q} = 2, Q_r = 2Q$ .

$$F = \frac{1 + 2}{(1 + 0.1(2))^2} = 2.08$$

$$85.4 = \frac{100}{1 + 0.448 \sqrt{\frac{5200}{V^2(2.08)}}} \quad \text{or} \quad 2.08V^2 = 0.1456 \quad \text{or} \quad V = 131.02 \text{ m}^3$$

Hydraulic loading rate  $\rightarrow (Q + Q_r)/A_c$  organic loading rate  $\rightarrow Q(\text{BOD})/V$ .

Assuming tank is circular  $\rightarrow \frac{\pi d^2}{4} \times \text{depth} = 131.02 \text{ m}^3$  (Assume tank is 2m deep).

$$d = 9.13 \text{ m (diameter)}$$

$$\text{Hydraulic loading rate} \rightarrow \frac{(Q + 2Q)}{(\frac{131.02}{2}) \text{ m}^2} = \frac{3(20000 \text{ m}^3/\text{d})}{65.51 \text{ m}^2} = 915.89 \text{ m}^3/\text{m}^2 \cdot \text{d}$$

$$\text{organic loading rate} \rightarrow \frac{20000 \text{ m}^3/\text{d} (260 \text{ mg/L})}{131.02 \text{ m}^3} = 39.68 \text{ kg/m}^3 \cdot \text{d}$$

